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5-Hydroxy-7-methoxy-2-methyl-4H-chromen-4-one from *Dysoxylum macrocarpum* (Meliaceae)Ibrahim A. Najmuldeen,^a Abdul Hamid Abdul Hadi,^a Khalijah Awang,^a Khalit Mohamad^b and Seik Weng Ng^{a*}^aDepartment of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia, and ^bDepartment of Pharmacy, Faculty of Medicine, University of Malaya, 50603 Kuala Lumpur, Malaysia

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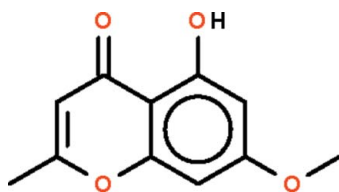
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Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.001$ Å; R factor = 0.035; wR factor = 0.108; data-to-parameter ratio = 15.1.

Both independent molecules in the asymmetric unit of the title compound, $\text{C}_{11}\text{H}_{10}\text{O}_4$, are almost planar (r.m.s. deviations = 0.011 and 0.033 Å). In both molecules, the hydroxy group is intramolecularly hydrogen bonded to the ketonic O atom. The independent molecules are stacked alternately along the a axis, with the centroids of their chromene ring separated by distances of 4.490 (1) and 3.621 (1) Å.

Related literature

For studies on other *Dysoxylum* species, see: Ismail *et al.* (2009); Lakshmi *et al.* (2007); Mohamad *et al.* (1999); Mohanakumara *et al.* (2010); Senthil Nathan *et al.* (2008); Xie *et al.* (2008).



Experimental

Crystal data

$\text{C}_{11}\text{H}_{10}\text{O}_4$
 $M_r = 206.19$
 Monoclinic, $P2_1/c$

$a = 7.7393$ (3) Å
 $b = 14.5373$ (6) Å
 $c = 16.8263$ (7) Å

$\beta = 98.848$ (1)°
 $V = 1870.57$ (13) Å³
 $Z = 8$
 Mo $K\alpha$ radiation

$\mu = 0.11$ mm⁻¹
 $T = 100$ K
 $0.35 \times 0.35 \times 0.02$ mm

Data collection

Bruker SMART APEXII area-detector diffractometer
 17796 measured reflections

4285 independent reflections
 3795 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.017$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.035$
 $wR(F^2) = 0.108$
 $S = 1.02$
 4285 reflections
 283 parameters
 2 restraints

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.34$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.28$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{O3}-\text{H3}\cdots\text{O2}$	0.85 (1)	1.83 (1)	2.618 (1)	154 (2)
$\text{O7}-\text{H7}\cdots\text{O6}$	0.85 (1)	1.79 (1)	2.595 (1)	156 (2)

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *pubCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CI5120).

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supplementary materials

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5-Hydroxy-7-methoxy-2-methyl-4*H*-chromen-4-one from *Dysoxylum macrocarpum* (Meliaceae)

I. A. Najmuldeen, A. H. Abdul Hadi, K. Awang, K. Mohamad and S. W. Ng

Comment

Dysoxylum produces terpenes, terpenoids and other compounds, as noted in studies on *Dysoxylum acutangulum*, *Dysoxylum beddomei*, *Dysoxylum binectariferum*, *Dysoxylum densiflorum*, *Dysoxylum malabaricum* and *Dysoxylum macranthum* (Ismail *et al.*, 2009; Lakshmi *et al.*, 2007; Mohamad *et al.*, 1999; Mohanakumara *et al.*, 2010; Senthil Nathan *et al.*, 2008; Xie *et al.*, 2008). A coumarin (Scheme I) is isolated from *Dysoxylum macrocarpum* in the present study. There are two independent molecules (Fig. 1). Both independent molecules are planar [r.m.s. deviations 0.011 and 0.033 Å]; one is stacked over the other [dihedral angle between the planes 4.5 (1) °] but the distance between them exceeds 3.5 Å. The hydroxy group is intramolecularly hydrogen bonded to the ketonic oxygen atom.

Experimental

Dried ground leaves of *Dysoxylum macrocarpum* (1.4 kg) were soaked in hexane for three days. The solvent was removed and the plant material was dried; the dried plant material was extracted with dichloromethane for another three days. The dichloromethane was removed by evaporation to give a crude material (30 g) that was subjected to column chromatography over silica gel. Separation was effected with hexane-dichloromethane (1:0 to 0:1 v/v); the polarity was increased with methanol. One fraction was eluted with hexane-methanol (92:8 v/v) to give a compound that was further purified by silica gel column chromatography with hexane-ethyl acetate (70:30 v/v) to yield colourless crystals.

Refinement

Carbon-bound H-atoms were placed in calculated positions (C–H = 0.95–0.98 Å) and were included in the refinement in the riding model approximation, with $U_{iso}(\text{H})$ set to 1.2 to 1.5 $U_{eq}(\text{C})$. The hydroxy H-atoms were located in a difference Fourier map, and were refined with a distance restraint of O–H = 0.84 (1) Å; their U_{iso} parameters were freely refined.

Figures

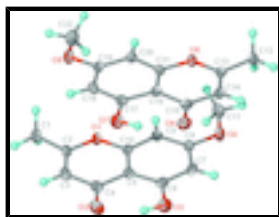


Fig. 1. Displacement ellipsoid plot (Barbour, 2001) of the two independent molecules of C₁₁H₁₀O₄ at the 70% probability level. H atoms are drawn as spheres of arbitrary radii.

5-Hydroxy-7-methoxy-2-methyl-4H-chromen-4-one

Crystal data

$C_{11}H_{10}O_4$	$F(000) = 864$
$M_r = 206.19$	$D_x = 1.464 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
Hall symbol: -P 2ybc	Cell parameters from 9622 reflections
$a = 7.7393 (3) \text{ \AA}$	$\theta = 2.5\text{--}28.3^\circ$
$b = 14.5373 (6) \text{ \AA}$	$\mu = 0.11 \text{ mm}^{-1}$
$c = 16.8263 (7) \text{ \AA}$	$T = 100 \text{ K}$
$\beta = 98.848 (1)^\circ$	Plate, colourless
$V = 1870.57 (13) \text{ \AA}^3$	$0.35 \times 0.35 \times 0.02 \text{ mm}$
$Z = 8$	

Data collection

Bruker SMART APEXII area-detector diffractometer	3795 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube graphite	$R_{\text{int}} = 0.017$
ω scans	$\theta_{\text{max}} = 27.5^\circ$, $\theta_{\text{min}} = 1.9^\circ$
17796 measured reflections	$h = -10 \rightarrow 10$
4285 independent reflections	$k = -18 \rightarrow 18$
	$l = -20 \rightarrow 21$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.035$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.108$	H atoms treated by a mixture of independent and constrained refinement
$S = 1.02$	$w = 1/[\sigma^2(F_o^2) + (0.0679P)^2 + 0.4625P]$
4285 reflections	where $P = (F_o^2 + 2F_c^2)/3$
283 parameters	$(\Delta/\sigma)_{\text{max}} = 0.001$
2 restraints	$\Delta\rho_{\text{max}} = 0.34 \text{ e \AA}^{-3}$
	$\Delta\rho_{\text{min}} = -0.28 \text{ e \AA}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.49963 (9)	0.31112 (5)	0.80755 (4)	0.01545 (16)
O2	0.54836 (10)	0.03128 (5)	0.81510 (5)	0.02311 (18)
O3	0.66721 (11)	0.03755 (5)	0.67769 (5)	0.02290 (18)

O4	0.71726 (10)	0.32572 (5)	0.55887 (4)	0.01882 (17)
O5	1.05826 (9)	0.38133 (5)	0.70570 (4)	0.01574 (16)
O6	1.12544 (10)	0.10446 (5)	0.73680 (5)	0.02192 (17)
O7	1.00123 (11)	0.11679 (5)	0.87114 (5)	0.02253 (18)
O8	0.82751 (10)	0.40629 (5)	0.95136 (4)	0.01974 (17)
C1	0.39489 (14)	0.32353 (7)	0.93075 (6)	0.0200 (2)
H1A	0.3576	0.2864	0.9737	0.030*
H1B	0.2964	0.3607	0.9049	0.030*
H1C	0.4909	0.3641	0.9536	0.030*
C2	0.45527 (12)	0.26180 (7)	0.87004 (6)	0.0163 (2)
C3	0.46768 (13)	0.16950 (7)	0.87405 (6)	0.0184 (2)
H3A	0.4329	0.1383	0.9186	0.022*
C4	0.53257 (12)	0.11694 (7)	0.81228 (6)	0.0172 (2)
C5	0.58040 (12)	0.17065 (6)	0.74648 (6)	0.0147 (2)
C6	0.64722 (12)	0.12963 (7)	0.68075 (6)	0.0160 (2)
C7	0.69215 (12)	0.18335 (7)	0.61950 (6)	0.0167 (2)
H7A	0.7379	0.1559	0.5759	0.020*
C8	0.66943 (12)	0.27902 (7)	0.62233 (6)	0.0148 (2)
C9	0.60437 (12)	0.32217 (6)	0.68517 (6)	0.01419 (19)
H9	0.5895	0.3870	0.6864	0.017*
C10	0.56218 (12)	0.26615 (6)	0.74610 (5)	0.01350 (19)
C11	0.69519 (14)	0.42341 (7)	0.55749 (6)	0.0205 (2)
H11A	0.7287	0.4483	0.5079	0.031*
H11B	0.7693	0.4506	0.6041	0.031*
H11C	0.5725	0.4383	0.5596	0.031*
C12	1.16596 (14)	0.38672 (7)	0.58240 (6)	0.0200 (2)
H12A	1.2153	0.3477	0.5440	0.030*
H12B	1.2512	0.4340	0.6033	0.030*
H12C	1.0590	0.4162	0.5553	0.030*
C13	1.12406 (12)	0.32926 (7)	0.65014 (6)	0.0163 (2)
C14	1.14769 (13)	0.23789 (7)	0.65825 (6)	0.0180 (2)
H14	1.1940	0.2047	0.6177	0.022*
C15	1.10416 (12)	0.18917 (7)	0.72722 (6)	0.0164 (2)
C16	1.03515 (12)	0.24579 (6)	0.78581 (6)	0.01414 (19)
C17	0.98490 (12)	0.20810 (6)	0.85677 (6)	0.0159 (2)
C18	0.91807 (12)	0.26380 (7)	0.91094 (6)	0.0167 (2)
H18	0.8858	0.2386	0.9586	0.020*
C19	0.89823 (12)	0.35822 (7)	0.89482 (6)	0.0151 (2)
C20	0.94688 (12)	0.39841 (6)	0.82650 (6)	0.01472 (19)
H20	0.9348	0.4626	0.8167	0.018*
C21	1.01372 (12)	0.34029 (6)	0.77342 (5)	0.01377 (19)
C22	0.78973 (16)	0.50156 (7)	0.93552 (7)	0.0250 (2)
H22A	0.7344	0.5275	0.9791	0.038*
H22B	0.7103	0.5077	0.8845	0.038*
H22C	0.8986	0.5347	0.9321	0.038*
H3	0.634 (2)	0.0179 (12)	0.7204 (8)	0.061 (6)*
H7	1.043 (3)	0.0968 (13)	0.8304 (9)	0.070 (6)*

supplementary materials

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0181 (3)	0.0157 (3)	0.0134 (3)	0.0013 (3)	0.0054 (3)	0.0001 (2)
O2	0.0254 (4)	0.0153 (4)	0.0293 (4)	0.0011 (3)	0.0063 (3)	0.0063 (3)
O3	0.0307 (4)	0.0125 (3)	0.0266 (4)	0.0021 (3)	0.0078 (3)	-0.0015 (3)
O4	0.0253 (4)	0.0179 (4)	0.0146 (3)	0.0024 (3)	0.0074 (3)	0.0023 (3)
O5	0.0205 (3)	0.0151 (3)	0.0125 (3)	-0.0002 (3)	0.0055 (3)	0.0011 (2)
O6	0.0276 (4)	0.0138 (3)	0.0246 (4)	0.0010 (3)	0.0045 (3)	-0.0030 (3)
O7	0.0336 (4)	0.0123 (3)	0.0226 (4)	0.0006 (3)	0.0071 (3)	0.0036 (3)
O8	0.0259 (4)	0.0179 (4)	0.0174 (4)	0.0017 (3)	0.0094 (3)	-0.0009 (3)
C1	0.0191 (5)	0.0263 (5)	0.0152 (5)	-0.0007 (4)	0.0048 (4)	-0.0019 (4)
C2	0.0131 (4)	0.0229 (5)	0.0129 (4)	-0.0009 (4)	0.0020 (3)	0.0014 (4)
C3	0.0175 (5)	0.0222 (5)	0.0157 (5)	-0.0006 (4)	0.0033 (4)	0.0049 (4)
C4	0.0141 (4)	0.0171 (5)	0.0198 (5)	-0.0005 (3)	0.0007 (3)	0.0032 (4)
C5	0.0135 (4)	0.0146 (4)	0.0157 (5)	0.0000 (3)	0.0013 (3)	0.0006 (3)
C6	0.0154 (4)	0.0139 (4)	0.0180 (5)	0.0009 (3)	0.0003 (3)	-0.0022 (3)
C7	0.0176 (5)	0.0173 (5)	0.0151 (4)	0.0018 (3)	0.0024 (3)	-0.0034 (3)
C8	0.0144 (4)	0.0182 (5)	0.0117 (4)	-0.0005 (3)	0.0014 (3)	0.0012 (3)
C9	0.0152 (4)	0.0126 (4)	0.0146 (4)	0.0004 (3)	0.0016 (3)	0.0008 (3)
C10	0.0127 (4)	0.0152 (4)	0.0127 (4)	0.0008 (3)	0.0020 (3)	-0.0014 (3)
C11	0.0251 (5)	0.0175 (5)	0.0200 (5)	0.0011 (4)	0.0064 (4)	0.0051 (4)
C12	0.0216 (5)	0.0232 (5)	0.0162 (5)	-0.0034 (4)	0.0059 (4)	0.0006 (4)
C13	0.0143 (4)	0.0209 (5)	0.0136 (4)	-0.0031 (3)	0.0023 (3)	-0.0024 (3)
C14	0.0178 (4)	0.0206 (5)	0.0164 (5)	-0.0015 (4)	0.0050 (4)	-0.0049 (4)
C15	0.0144 (4)	0.0157 (4)	0.0186 (5)	-0.0011 (3)	0.0007 (3)	-0.0030 (4)
C16	0.0140 (4)	0.0137 (4)	0.0143 (4)	-0.0010 (3)	0.0010 (3)	-0.0002 (3)
C17	0.0165 (4)	0.0137 (4)	0.0170 (5)	-0.0016 (3)	0.0008 (3)	0.0021 (3)
C18	0.0175 (5)	0.0178 (5)	0.0151 (4)	-0.0023 (4)	0.0029 (3)	0.0032 (3)
C19	0.0142 (4)	0.0175 (5)	0.0139 (4)	-0.0007 (3)	0.0030 (3)	-0.0022 (3)
C20	0.0164 (4)	0.0126 (4)	0.0152 (4)	0.0000 (3)	0.0025 (3)	0.0004 (3)
C21	0.0135 (4)	0.0156 (4)	0.0120 (4)	-0.0020 (3)	0.0014 (3)	0.0018 (3)
C22	0.0343 (6)	0.0178 (5)	0.0250 (5)	0.0052 (4)	0.0110 (4)	-0.0019 (4)

Geometric parameters (\AA , $^\circ$)

O1—C2	1.3596 (11)	C7—H7A	0.95
O1—C10	1.3729 (11)	C8—C9	1.3888 (13)
O2—C4	1.2514 (12)	C9—C10	1.3870 (13)
O3—C6	1.3494 (11)	C9—H9	0.95
O3—H3	0.850 (9)	C11—H11A	0.98
O4—C8	1.3637 (11)	C11—H11B	0.98
O4—C11	1.4302 (12)	C11—H11C	0.98
O5—C13	1.3611 (11)	C12—C13	1.4886 (13)
O5—C21	1.3761 (11)	C12—H12A	0.98
O6—C15	1.2496 (12)	C12—H12B	0.98
O7—C17	1.3517 (11)	C12—H12C	0.98
O7—H7	0.852 (9)	C13—C14	1.3450 (14)

O8—C19	1.3612 (11)	C14—C15	1.4430 (14)
O8—C22	1.4319 (12)	C14—H14	0.95
C1—C2	1.4880 (13)	C15—C16	1.4480 (13)
C1—H1A	0.98	C16—C21	1.3956 (13)
C1—H1B	0.98	C16—C17	1.4212 (13)
C1—H1C	0.98	C17—C18	1.3780 (14)
C2—C3	1.3463 (14)	C18—C19	1.4030 (13)
C3—C4	1.4412 (14)	C18—H18	0.95
C3—H3A	0.95	C19—C20	1.3920 (13)
C4—C5	1.4486 (13)	C20—C21	1.3862 (13)
C5—C10	1.3955 (13)	C20—H20	0.95
C5—C6	1.4220 (13)	C22—H22A	0.98
C6—C7	1.3795 (14)	C22—H22B	0.98
C7—C8	1.4035 (13)	C22—H22C	0.98
C2—O1—C10	119.45 (8)	O4—C11—H11C	109.5
C6—O3—H3	104.5 (13)	H11A—C11—H11C	109.5
C8—O4—C11	117.45 (7)	H11B—C11—H11C	109.5
C13—O5—C21	119.70 (8)	C13—C12—H12A	109.5
C17—O7—H7	103.2 (14)	C13—C12—H12B	109.5
C19—O8—C22	117.32 (8)	H12A—C12—H12B	109.5
C2—C1—H1A	109.5	C13—C12—H12C	109.5
C2—C1—H1B	109.5	H12A—C12—H12C	109.5
H1A—C1—H1B	109.5	H12B—C12—H12C	109.5
C2—C1—H1C	109.5	C14—C13—O5	122.61 (9)
H1A—C1—H1C	109.5	C14—C13—C12	126.13 (9)
H1B—C1—H1C	109.5	O5—C13—C12	111.25 (8)
C3—C2—O1	122.67 (9)	C13—C14—C15	121.42 (9)
C3—C2—C1	126.45 (9)	C13—C14—H14	119.3
O1—C2—C1	110.87 (8)	C15—C14—H14	119.3
C2—C3—C4	121.49 (9)	O6—C15—C14	123.03 (9)
C2—C3—H3A	119.3	O6—C15—C16	121.83 (9)
C4—C3—H3A	119.3	C14—C15—C16	115.14 (9)
O2—C4—C3	122.89 (9)	C21—C16—C17	117.60 (8)
O2—C4—C5	122.03 (9)	C21—C16—C15	120.37 (9)
C3—C4—C5	115.08 (9)	C17—C16—C15	122.02 (9)
C10—C5—C6	117.53 (8)	O7—C17—C18	119.54 (9)
C10—C5—C4	120.14 (9)	O7—C17—C16	119.91 (9)
C6—C5—C4	122.32 (9)	C18—C17—C16	120.55 (9)
O3—C6—C7	119.31 (9)	C17—C18—C19	119.21 (9)
O3—C6—C5	120.21 (9)	C17—C18—H18	120.4
C7—C6—C5	120.48 (9)	C19—C18—H18	120.4
C6—C7—C8	119.23 (9)	O8—C19—C20	123.40 (9)
C6—C7—H7A	120.4	O8—C19—C18	114.27 (8)
C8—C7—H7A	120.4	C20—C19—C18	122.33 (9)
O4—C8—C9	123.04 (8)	C21—C20—C19	116.84 (8)
O4—C8—C7	114.59 (8)	C21—C20—H20	121.6
C9—C8—C7	122.37 (9)	C19—C20—H20	121.6
C10—C9—C8	116.85 (8)	O5—C21—C20	115.79 (8)
C10—C9—H9	121.6	O5—C21—C16	120.75 (8)

supplementary materials

C8—C9—H9	121.6	C20—C21—C16	123.46 (8)
O1—C10—C9	115.31 (8)	O8—C22—H22A	109.5
O1—C10—C5	121.16 (8)	O8—C22—H22B	109.5
C9—C10—C5	123.53 (8)	H22A—C22—H22B	109.5
O4—C11—H11A	109.5	O8—C22—H22C	109.5
O4—C11—H11B	109.5	H22A—C22—H22C	109.5
H11A—C11—H11B	109.5	H22B—C22—H22C	109.5
C10—O1—C2—C3	-0.95 (13)	C21—O5—C13—C14	0.18 (14)
C10—O1—C2—C1	178.64 (7)	C21—O5—C13—C12	-179.19 (7)
O1—C2—C3—C4	1.16 (15)	O5—C13—C14—C15	-0.03 (15)
C1—C2—C3—C4	-178.35 (9)	C12—C13—C14—C15	179.24 (9)
C2—C3—C4—O2	178.74 (10)	C13—C14—C15—O6	-179.58 (9)
C2—C3—C4—C5	-0.66 (14)	C13—C14—C15—C16	-0.25 (14)
O2—C4—C5—C10	-179.40 (9)	O6—C15—C16—C21	179.72 (9)
C3—C4—C5—C10	0.00 (13)	C14—C15—C16—C21	0.38 (13)
O2—C4—C5—C6	0.38 (15)	O6—C15—C16—C17	-1.11 (15)
C3—C4—C5—C6	179.78 (8)	C14—C15—C16—C17	179.55 (8)
C10—C5—C6—O3	-179.79 (8)	C21—C16—C17—O7	179.27 (8)
C4—C5—C6—O3	0.42 (14)	C15—C16—C17—O7	0.08 (14)
C10—C5—C6—C7	0.16 (14)	C21—C16—C17—C18	-0.18 (14)
C4—C5—C6—C7	-179.63 (8)	C15—C16—C17—C18	-179.37 (8)
O3—C6—C7—C8	179.29 (9)	O7—C17—C18—C19	-178.73 (9)
C5—C6—C7—C8	-0.66 (14)	C16—C17—C18—C19	0.72 (14)
C11—O4—C8—C9	-1.20 (13)	C22—O8—C19—C20	5.00 (14)
C11—O4—C8—C7	179.17 (8)	C22—O8—C19—C18	-174.86 (9)
C6—C7—C8—O4	-179.82 (8)	C17—C18—C19—O8	178.59 (8)
C6—C7—C8—C9	0.55 (14)	C17—C18—C19—C20	-1.27 (14)
O4—C8—C9—C10	-179.53 (8)	O8—C19—C20—C21	-178.65 (8)
C7—C8—C9—C10	0.08 (14)	C18—C19—C20—C21	1.19 (14)
C2—O1—C10—C9	-179.84 (8)	C13—O5—C21—C20	-179.66 (8)
C2—O1—C10—C5	0.26 (13)	C13—O5—C21—C16	-0.03 (13)
C8—C9—C10—O1	179.48 (7)	C19—C20—C21—O5	178.99 (8)
C8—C9—C10—C5	-0.61 (14)	C19—C20—C21—C16	-0.62 (14)
C6—C5—C10—O1	-179.60 (8)	C17—C16—C21—O5	-179.46 (8)
C4—C5—C10—O1	0.19 (14)	C15—C16—C21—O5	-0.26 (14)
C6—C5—C10—C9	0.50 (14)	C17—C16—C21—C20	0.13 (14)
C4—C5—C10—C9	-179.71 (9)	C15—C16—C21—C20	179.34 (9)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
O3—H3 \cdots O2	0.85 (1)	1.83 (1)	2.618 (1)	154 (2)
O7—H7 \cdots O6	0.85 (1)	1.79 (1)	2.595 (1)	156 (2)
C7—H7A \cdots O8 ⁱ	0.95	2.48	3.4222 (12)	173
C9—H9 \cdots O2 ⁱⁱ	0.95	2.35	3.2614 (12)	160
C12—H12B \cdots O2 ⁱⁱⁱ	0.98	2.37	3.3326 (13)	166
C18—H18 \cdots O4 ^{iv}	0.95	2.47	3.3904 (12)	163
C20—H20 \cdots O6 ⁱⁱⁱ	0.95	2.27	3.1995 (12)	166

Symmetry codes: (i) $x, -y+1/2, z-1/2$; (ii) $-x+1, y+1/2, -z+3/2$; (iii) $-x+2, y+1/2, -z+3/2$; (iv) $x, -y+1/2, z+1/2$.

Fig. 1

